

ror surface can be calculated and the "high" or "low" spots, lying outside permissible accuracy can be ascertained.

It is possible to obtain accuracy only  $1/100$  the wavelength of light by the method, which is not only applicable to parabolic mirrors but to optical surfaces

which depart considerably from a spherical shape. These forms, Dr. Gaviola states, have been avoided up to now for it is a rule of optical shops that "a surface can be figured as accurately and only as accurately as it can be tested."

*Science News Letter, December 2, 1939*

PALEONTOLOGY

## Coal Mine Waste Piles Yield Beautiful Fossil Leaves

See Front Cover

**B**EAUTY is sought and found in about the unlikeliest of all imaginable places—the waste-heaps of soft coal mines—by George Langford of Joliet, Ill. For him it is not a profession but an absorbing avocation, for Mr. Langford makes his living as manager of a manufacturing plant in Joliet. His spare time he spends botanizing the forests of 250 million years ago, as represented by the fossil leaf-remains embedded in the nodules of hardened clay cast out with the shaly waste from the coal pits of central Illinois.

Finding fine leaf-imprints in shale nodules is not particularly new in itself. Thousands of them are in geological museums and private collections everywhere. The real contribution of Mr. Langford is the method he has developed for bringing out their full color and beauty, something not accomplished by the kind of handling hitherto employed. Leaf-forms that ordinarily show dull and colorless take on almost the sheen of living flowers.

A shale nodule is a more or less elongated oval lump of stuff that looks like hardened clay. Indeed it *is* hardened clay, solidified around a leaf, or twig, or insect, or other object that fell into the mud and became embedded there, ages ago when coal was in the making. If you turn it edgewise and give it a sharp rap with a hammer it will split apart, showing the convex or positive side of the leaf-print on the face of one of the halves and the concave or negative side on the other.

That is all that has been done with leaf-nodules hitherto—just split them open to show the contents. But in that state the leaf-prints are dull and rather uninteresting, due to their being covered with tightly sticking tiny mineral particles—literally the dust of ages.

Mr. Langford undertook to get rid

of this millennial dust. His technique is something between a jeweler's and a dentist's. He uses the small, keenly pointed and edged tools of a dentist, plus a variety of brushes, to dislodge the mineral particles, and then carefully burnishes each small detail of the cleaned-up design by hand. No buffing wheel, however fine, is permitted to touch one of his precious specimens.

Most people, when they first see a Langford leaf-nodule preparation, imagine that it has been painted or enameled. This is not the case. No artificial color of any kind is added, no varnish, no shellac. A little colorless burnishing fluid is the only aid he employs. The colors are those latent in the minerals composing the specimen itself.

Mr. Langford states: "The stone of the strip mine nodules is very variable. All of it contains more or less iron. In some places it is a buff-colored and rather soft shale. Occasionally it is sandstone. The stone best suited for fossil preservation is very close-grained and hard, in varying shades of fawn, gray, light blue and dark blue.

"Weathering converts the iron content into oxide, producing red, which, combining with the stone colors, results in orange, brown, purple and other colors. It remained for us to develop these colors by accentuating them in the plants and suppressing them in the surrounding stone. The result is a fine reproduction of the plant in color, and of surprisingly lifelike appearance."

The plant remains brought out by Mr. Langford's method are almost all leaves of ferns and fern-like plants, which were the highest forms of vegetation on the earth during the coal age. Higher flowering plants did not appear until much later. It is in the leaves, therefore, their delicate venation and smooth surface texture, that the greatest possible development of beauty must be sought.

*Science News Letter, December 2, 1939*

PHYSICS

## "Magnetic" Mines Feasible, Say American Experts

**F**LOATING mines exploded by the changes in magnetic field, due to the near presence of an iron ship's hull, may be the new factor in renewed attacks by Germany against Allied and neutral shipping.

Accepting British press stories of "magnetic" mines as plausible, government and naval scientists in the United States privately speculated on how they work. Without much difficulty, they believe, it should be possible to create a firing mechanism for floating mines that would be set off by slight changes in magnetic field as an iron-hulled ship passed close by. Only when the ship was close—and hence vulnerable—would these mines go off. There would be no explosion at farther distances at which the mines would be harmless. And yet they would not have to make actual contact with the ship as do ordinary mines.

Dropping small mines from an airplane, as reported, is perfectly feasible and would be no different than dropping torpedoes from planes, as is now done. The only thing required would be explosives which are not set off by a jarring impact.

Press reports of magnetic mines that would be attracted directly to the iron hull of a ship with Machiavellian deadliness can be ruled out, say the scientists, unless Nazi ingenuity has succeeded in circumventing the rigid, fixed and well-known laws of magnetic attraction.

Basic trouble with this scheme is that magnetic attraction decreases inversely as the square of the distance so that any magnetic force at ten feet is only one one-hundredth of that at one foot. At twenty feet the attractive force would be only one four-hundredth, and so on.

It is pointed out that even if very intense magnetic fields could be created in a mine by highly magnetic alloys, or possibly by a compact electromagnet inside the mine, the mere presence of this potent field would make it extremely easy to detect such mines lying on the bottom or floating beneath the surface. There are many detecting devices, such as those used in geophysical prospecting, which could do this.

The idea of a magnetic firing mechanism is something else, however, and might explain the sinking of mine sweepers recently. Their iron hulls could set off such magnetic mines under the hull before the "sweeps" could gather them.

*Science News Letter, December 2, 1939*